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## Key indicators

Single-crystal X-ray study  
*T* = 288 K  
Mean  $\sigma(\text{C}-\text{C}) = 0.004 \text{ \AA}$   
*R* factor = 0.039  
*wR* factor = 0.104  
Data-to-parameter ratio = 14.8For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.1,1'-(1,2-Ethanediy)bis(2,3-dihydro-3-methyl-1*H*-imidazole-2-thione)The title compound,  $\text{C}_{10}\text{H}_{14}\text{N}_4\text{S}_2$ , obtained from the alkylation of methimazole with ethylene dibromide, is centrosymmetric and the two imidazole-2-thione ring moieties are anti-parallel to one another.

## Comment

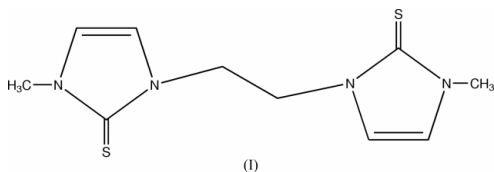
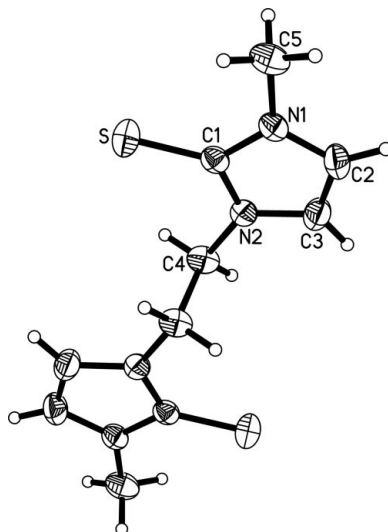
Acyclic crown ethers with nitrogen heterocycles as the end groups, which can coordinate and transport metals, continue to receive attention (Vögtle & Weber, 1979; Meth-Cohn & Smith, 1982; Liu & Shi, 1992; Liu *et al.*, 1992, 2001; Matthews *et al.*, 1996). We have reported the syntheses, characterization and coordination properties of non-cyclic crown ethers with methimazole end groups (Liu & Shi, 1992; Liu *et al.*, 1992). We report here the X-ray crystal structure of the title compound, (I).The whole molecule possesses a centre of symmetry (Fig. 1). The two imidazole-2-thione ring moieties are anti-parallel to one another. The perpendicular distance between the two rings is 1.504 (2) Å. The bridging chain moiety,  $-\text{CH}_2\text{CH}_2-$ , adopts an antiperiplanar conformation. The bond lengths and angles in (I) have normal values (Table 1). There is no intermolecular  $\text{S} \cdots \text{S}$  short contact in the crystal packing (Fig. 2).

Figure 1

The molecular structure of (I), showing 50% probability displacement ellipsoids, and the numbering scheme for non-H atoms.

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Professor Mingli Shi on the  
occasion of his 70 th birthday.

## Experimental

The title compound, (I), was prepared *via* alkylation of methimazole with ethylene dibromide under phase-transfer catalytic conditions (m.p. 467–468.5 K). Single crystals of (I) suitable for X-ray diffraction analysis were obtained by slow evaporation of an ethanol solution.

### Crystal data

$C_{10}H_{14}N_4S_2$	$D_x = 1.365 \text{ Mg m}^{-3}$
$M_r = 254.37$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 27 reflections
$a = 4.667 (1) \text{ \AA}$	$\theta = 4.8\text{--}13.6^\circ$
$b = 19.720 (3) \text{ \AA}$	$\mu = 0.41 \text{ mm}^{-1}$
$c = 7.037 (1) \text{ \AA}$	$T = 288 (2) \text{ K}$
$\beta = 107.08 (1)^\circ$	Block, colourless
$V = 619.0 (2) \text{ \AA}^3$	$0.40 \times 0.28 \times 0.20 \text{ mm}$
$Z = 2$	

### Data collection

Siemens P4 diffractometer	$R_{\text{int}} = 0.029$
$\omega$ scans	$\theta_{\text{max}} = 25.0^\circ$
Absorption correction: refined from $\Delta F$ (SHELXA; Sheldrick, 1997)	$h = 0 \rightarrow 5$
$T_{\text{min}} = 0.860$ , $T_{\text{max}} = 0.921$	$k = 0 \rightarrow 23$
1339 measured reflections	$l = -8 \rightarrow 8$
1097 independent reflections	3 standard reflections every 97 reflections
804 reflections with $I > 2\sigma(I)$	intensity decay: 2.9%

### Refinement

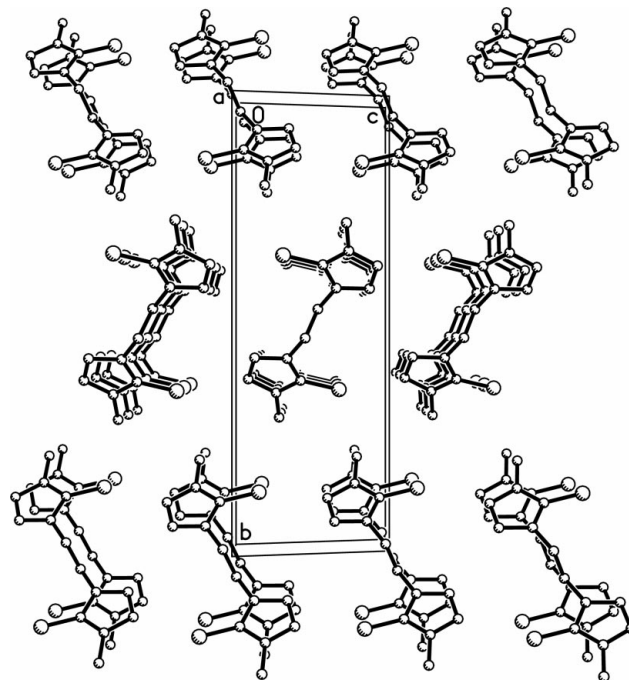
Refinement on $F^2$	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.039$	$w = 1/[\sigma^2(F_o^2) + (0.0578P)^2]$
$wR(F^2) = 0.104$	where $P = (F_o^2 + 2F_c^2)/3$
$S = 1.01$	$(\Delta/\sigma)_{\text{max}} < 0.001$
1097 reflections	$\Delta\rho_{\text{max}} = 0.22 \text{ e \AA}^{-3}$
74 parameters	$\Delta\rho_{\text{min}} = -0.21 \text{ e \AA}^{-3}$

**Table 1**

Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ ).

S—C1	1.679 (2)	N2—C3	1.381 (3)
N1—C1	1.358 (3)	N2—C4	1.451 (3)
N1—C2	1.374 (3)	C2—C3	1.329 (4)
N2—C1	1.364 (3)		
C1—N1—C5	124.6 (2)	N1—C1—S	127.63 (19)
C1—N2—C4	124.4 (2)	N2—C1—S	127.16 (18)
C3—N2—C4	125.9 (2)		
C5—N1—C1—N2	177.4 (2)	C3—N2—C1—S	−178.99 (19)
C2—N1—C1—S	179.3 (2)	C4—N2—C1—S	−1.1 (3)
C5—N1—C1—S	−3.6 (3)	C1—N1—C2—C3	−0.4 (3)
C4—N2—C1—N1	177.93 (19)	C4—N2—C3—C2	−178.2 (2)

All H atoms were positioned geometrically and treated as riding atoms, with C—H distances of 0.93 or 0.97  $\text{\AA}$ .



**Figure 2**

The molecular packing diagram of (I); H atoms are omitted.

Data collection: XSCANS (Siemens, 1994); cell refinement: XSCANS; data reduction: SHELXTL (Sheldrick, 1997); program(s) used to solve structure: SHELXTL; program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

## References

- Liu, Q. J., Li, L. Z., Liu, K. G., Qi, C. S., Gao, D. T., Li, L. H., Ma, C. L., Pan, F. M., Huang, R. Q. & Qu, R. J. (2001). *Chin. J. Org. Chem.* **21**, 160–162.
- Liu, Q. J. & Shi, M. L. (1992). *Chin. J. Org. Chem.* **12**, 509–513.
- Liu, Q. J., Shi, M. L., Jiang, C. Q. & Liu, F. L. (1992). *Chem. J. Chin. Univ.* **13**, 328–331.
- Matthews, C. J., Clegg, W., Elsegood, M. R. J., Leese, T. A., Thorp, D., Thornton, P. & Lockhart, J. C. (1996). *J. Chem. Soc. Dalton Trans.* pp. 1531–1538.
- Meth-Cohn, O. & Smith, D. I. (1982). *J. Chem. Soc. Perkin Trans. 1*, pp. 261–270.
- Sheldrick, G. M. (1997). *SHELXA and SHELXTL* (Version 5.1). Bruker AXS Inc., Madison, Wisconsin, USA.
- Siemens (1994). *XSCANS*. Version 2.1. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Vögtle, F. & Weber, E. (1979). *Angew. Chem. Int. Ed.* **18**, 753–776.